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Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.003 Å R factor = 0.020 wR factor = 0.048 Data-to-parameter ratio = 13.6

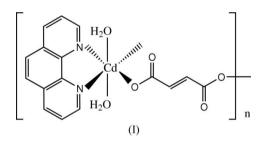
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

catena-Poly[[diaqua(1,10-phenanthroline- $\kappa^2 N, N'$)cadmium(II)]- μ -fumarato- $\kappa^2 O:O'$]

In the title polymeric complex, $[Cd(C_4H_2O_4)(C_{12}H_8N_2)-(H_2O)_2]_n$, the Cd^{II} ions are bridged by fumarate dianions to form zigzag complex chains. $\pi-\pi$ stacking is observed between the parallel phenanthroline ligands of neighboring complex chains. The Cd ion is located on a twofold axis and has a distorted octahedral coordination geometry, while the fumarate dianion lies on an inversion center.

Comment

Polynuclear complexes have received much attention in the past decade. Some polynuclear Cd^{II} complexes have shown photoluminescent properties (Tao *et al.*, 2000). We prepared the title Cd^{II} complex, (I), which displays a polymeric structure.



A segment of the polymeric structure of (I) is shown in Fig. 1. The Cd ion is located on a twofold axis and displays a distorted octahedral coordination geometry (Table 1), formed by two fumarate dianions, a 1,10-phenanthroline (phen) and two water molecules. The fumarate dianions, located on

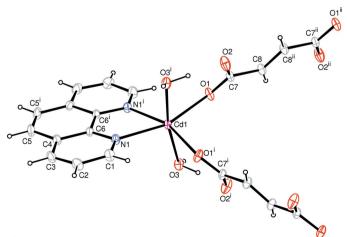
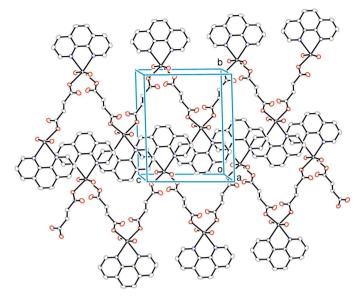


Figure 1

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved A segment of the polymeric structure of (I), with 30% probability displacement ellipsoids (arbitrary spheres for H atoms) [symmetry codes: (i) 2 - x, y, $-z + \frac{1}{2}$; (ii) $-x + \frac{5}{2}$, $-y + \frac{3}{2}$, 1 - z].

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A packing diagram for (I), showing π - π stacking between phen ligands. H atoms have been omitted.

inversion centers, bridge the neighboring Cd^{II} ions to form zigzag polymeric chains. The polymeric chains are arranged in such a way that the phen ligands of neighboring chains overlap (Fig. 2). The centroid-to-centroid separation of 3.693 (3) Å between N1-containing pyridine rings of parallel phen ligands suggests the existence of π - π stacking in the crystal structure of (I).

Experimental

An aqueous solution (30 ml) containing $Cd(NO_3)_2$ (0.1 mmol), fumaric acid (0.1 mmol), Na_2CO_3 (0.2 mmol) and phen (0.1 mmol) was refluxed for 2 h. After cooling to room temperature the solution was filtered. Single crystals of (I) were obtained from the filtrate after 3 d.

Crystal data

$[Cd(C_4H_2O_4)(C_{12}H_8N_2)(H_2O)_2]$
$M_r = 442.69$
Monoclinic, $C2/c$
a = 11.760 (9) Å
b = 13.06 (1) Å
c = 11.243 (9) Å
$\beta = 114.92 \ (1)^{\circ}$
V = 1566 (2) Å ³
Z = 4

 $D_x = 1.878 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 874 reflections $\theta = 2.5-26.0^{\circ}$ $\mu = 1.43 \text{ mm}^{-1}$ T = 293 (2) K Block, yellow $0.32 \times 0.28 \times 0.24 \text{ mm}$

Data collection

Bruker SMART CCD area-detector	1554 independent reflections
diffractometer	1416 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{int} = 0.023$
Absorption correction: multi-scan	$\theta_{max} = 26.2^{\circ}$
(<i>SADABS</i> ; Sheldrick, 1996)	$h = -11 \rightarrow 14$
$T_{min} = 0.628, T_{max} = 0.710$	$k = -16 \rightarrow 10$
4332 measured reflections	$l = -13 \rightarrow 13$
Refinement Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.020$ $wR(F^2) = 0.048$ S = 1.04 1554 reflections 114 parameters	H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0278P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.003$ $\Delta\rho_{\text{max}} = 0.27 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{\text{min}} = -0.37 \text{ e} \text{ Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Cd1-O1 Cd1-O3	2.3095 (19) 2.311 (2)	Cd1-N1	2.356 (2)
$O1-Cd1-O1^{i}$ O1-Cd1-O3 $O1-Cd1-O3^{i}$ O1-Cd1-N1	99.78 (10) 87.21 (7) 86.50 (7) 161.80 (6)	$O1^{i}-Cd1-N1$ $O3-Cd1-N1^{i}$ O3-Cd1-N1 $O3-Cd1-O3^{i}$	95.65 (8) 84.66 (6) 103.40 (6) 170.22 (9)

Symmetry code: (i) $-x + 2, y, -z + \frac{1}{2}$.

Water H atoms were located in a difference Fourier map and refined as riding, with $U_{\rm iso}({\rm H}) = 1.5 U_{\rm eq}({\rm O})$. Other H atoms were placed in calculated positions, with C–H = 0.93 Å and refined in riding mode with $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

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References

Bruker (2000). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Winsconsin, USA.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Tao, J., Tong, J.-X., Shi, J.-X., Chen, X.-M. & Ng, S. W. (2000). *Chem. Commun.* pp. 2043–2044.